

## SYLLABUS

| BOTANY | ZOOLOGY | PHYSICS | CHEMISTRY |
| :---: | :---: | :---: | :---: |
| Biological Classification | Neural Control and Co-ordination | Thermodynamics, <br> Kinetic Theory | Organic Chemistry - Some Basic <br> Principles and Techniques |



XI STD syllabus

## IMPORTANT INSTRUCTIONS:

1. There are four parts in the question paper A, B, C and D consisting of Botany, Zoology, Physics and Chemistry having $\mathbf{4 5}$ questions in each part of equal weightage. Each question is allotted $\mathbf{4}$ (four) marks for each correct response.
2. Candidates will be awarded marks as stated above in instruction for correct response of each question. $\mathbf{1 / 4}$ (one fourth) marks will be deducted for indicating incorrect response of each question. No deduction from the total score will be made if no response is indicated for an item in the answer sheet.
3. There is only one correct response for each question. Filling up more than one response in each question will be treated as wrong response and marks for wrong response will be deducted accordingly as per instruction.

## BOTANY

1. During Gram's stain
a) All bacteria whether Gram (+) ve or (-ve), take crystal violet stain
b) Only gram +ve bacteria take crystal violet stain
c) Only Gram - ve bacteria take crystal violet stain
d) Gram $(+)$ ve bacteria lose this stain after alcohol treatment and take red stain of safranin
2. Under optimum condition Bacterial cells divide once in $\mathbf{2 0}$ minutes by binary fission. How many bacteria will be produced in $\mathbf{2}$ hours with same rate of division ?
a) 8
b) 32
c) 128
d) 64
3. The archaebacteria occurring in marshes, swamps, rumens of cattles, gobar gas plants are
a) methanogens
b) ammonifying bacteria
c) thermoacidophiles
d) denitrifying bacteria
4. A symbiotic nitrogen fixing moneran among the following is
a) Nitrocystis
b) Anabaena
c) Nitrobacter
d) Escherichia
5. From where you will collect E. coli ?
a) Human excreta
b) On leaves
c) Water
d) Human stomach
6. A bacterial cell divides every minute. It takes one hour to fill a cup. Time required to fill half cup is
a) 60 minutes
b) 59 minutes
c) 30 minutes
d) 29 minutes
7. The joker of plant kingdom area
a) Bacteria
b) Archaebacteria
c) PPLO
d) Viriods
8. Which one of the following statements is wrong
a) Golden algae are also called desmids
b) Eubacteria are also called false bacteria
c) Phycomycetes are fungi
d) Cyanobacteria are also called blue - green alage
9. Select the wrong statement
a) Bacterial cell wall is made up of peptidoglycan
b) Pili and fimbriae are mainly involved in motility of bacterial cells
c) Cyanobacteria lack flagellated cells
d) Mycoplasma is a wall - less microorganism
10. Which one of the following is a slime mould
a) Thiobacillus
b) Anabaena
c) Rhizopus
d) Physarum
11. The 'fire' algae responsible for red tides are the red dinoflagellates, which are
a) Ceratium
b) Gonyaulax
c) Gymnodinium
d) (2) and (3)
12. Auxospores or rejuvenescent cells are characteristic of which of the followings
a) Dinoflagellates
b) Diatoms
c) Zooflagellates
d) Sporozoans
13. Paralytic shellfish poisoning (PSP) is caused by toxin saxitonin by
a) Vorticella
b) Ephidicum
c) Gonyaulax
d) Ceratium
14. Which animalcule is immortal?
a) Paramecium
b) Plasmodium
c) Amoeba
d) Euglena
15. Match the following and select the correct combination from the options given below

| Column I <br> (kingdom) |  | Column II (Class) |
| :--- | :--- | :--- |
| A. | Plantae | 1. Archaebacteria |
| B. | Fungi | 2. Euglenoids |
| C. | Protista | 3. Phycomycetes |
| D. | Monera | 4. Algae |

a) $\mathrm{A}-4, \mathrm{~B}-3, \mathrm{C}-2$, $\mathrm{D}-$
b) A-1, B- 2, C-3, D-4
c) A-3, B-4, C-2, D-1
d) A-2, B- $3, C-4, D-1$
16. Which of the following does not belong to the kingdom Prostia
a) Chrysophytes
b) Euglenoids
c) Ascomycetes
d) Dinoflagellates
17. In the five kingdom classification, chlamydomonas and chlorella have been included in
a) Protista
b) algae
c) Plantae
d) monera
18. The beautiful diatoms and desmids are placed under
a) chrysophytes
b) dinoflagellates
c) euglenoids
d) slime moulds
19. In which group of organisms the cell walls form two thin overlapping shells which fit together
a) Chrysophytes
b) Euglenoids
c) Dinoflagellates
d) Slime moulds
20. Chrysophytes, Euglenoids, Dinoflagellates and Slime moulds are included in the kingdom
a) Protista
b) Fungi
c) Animalia
d) Monera
21. Select the wrong statement
a) The walls or diatoms are easily destructible
b) 'Diatomaceous earth' is formed by the cell walls of diatoms
c) Diatoms are chief producers in the oceans
d) Diatoms are microscopic and float passively in water
22. Mycorrhiza is
a) symbiotic association of a soil fungus and roots of higher plants
b) Parasite association between a fungus and roots of seeded plants
c) saprophytic association between a fungus and roots of seeded plants
d) symbiotic association between an algae and fungi
23. In sac fungi (ascomycetes) the ascospores occur in sac like body known as
a) ascus
b) basidium
c) ascocarp
d) basidiocarp
24. The name Club fungi is given to basidiomycetes due to the presence of
a) Club shaped basidia
b) club shaped basidiospores
c) hymenium of basidia
d) water droplet mechanism for dehiscence of basidiospores
25. All members of fungi imperfecti (Deutero-mycetes) group lack
a) sexual reproduction
b) spores
c) asexual reproduction
d) hyphae
26. Rice crop was destroyed by a fungus which resulted in Bengal famine (1942-43). It was due to
a) Xanthomonas malvacearum b) Pyricularia oryzae c) Helminthosporium oryzea d) Puccinia graminis
27. A fungus whose extract of sclerotia can be chemically altered to produce powerful hallucinogenic drug (LSD) is
a) Claviceps purpurea
b) Aspergillus flavus
c) Amanita caesarea
d) Psilocybe mexicana
28. Meiosis occurs in the life cycle of Rhizopus during
a) formation of gemetangium
b) germination of zygospore
c) formation of germ sporangium
d) formation of aplanospores
29. Ascus in ascomycetes contains 8 ascospores. These are formed as a result of
a) two meiosis
b) one meiosis and one mitosis c
c) four mitosis
d) one meiosis and two mitosis
30. Champ connections occur in
a) Haplomycetes
b) Saccharomycetes
c) Basidiomycetes
d) Ascomycetes
31. The fruiting body formed from a filamentous heterotrophic organism which is known for its nutritive value for the humanity is
a) Cremocarp
b) Acervulus
c) Basidiocarp
d) Akinete
32. Find the correct match

| Name | Feature | Class |
| :--- | :--- | :--- |
| (a) Aspergillus | Aseptate mycelium | Ascomycetes |
| (b) Trichoderma | Imperfect fungi | Phycomycetes |
| (c) Rhizopus | Coenocytic mycelium | Deuteromycetes |
| (d) Puccinia | Branched and septate mycelium | Basidiomycetes |

33. Match the columns

| (A) | Early Blight of potato | (i) | Puccinia graminis |
| :--- | :--- | :--- | :--- |
| (B) | Late Blight to potato | (ii) | Ustilago tritici |
| (C) | Smut of Wheat | (iii) | Phytophthora infestans |
| (D) | Rust of Wheat | (iv) | Alternaria solani |

a) (A)-(i), (B) - (ii), (C) - (iii), (D) - (iv)
b) (A) - (iv), (B) - (iii), (C) - (ii), (D) - (i)
c) $(\mathrm{A})-(\mathrm{ii}),(\mathrm{B})-(\mathrm{iii}),(\mathrm{C})-(\mathrm{iv}),(\mathrm{D})-$ (i)
d) $(\mathrm{A})-$ (iv), (B) - (ii), (C) - (iii), (D) - (i)
34. Find the correct match
a) Phytophthora infestans - Early blight of potato
b) Albugo candida - white rust of crucifers
c) Puccinia graminis - Loose smut of wheat
d) All of the above
35. Which one of the following is endogenously produced
a) Ascospores
b) Basidiospores
c) Conidiospores
d) All of the above
36. The staple crop in Ireland was completely destroyed in 1844 - 1846 , by a fungal disease resulting in a great famine. The causal organism in this case was
a) Puccinia graminis
b) Ustilago tritici
c) Phytophthora infestans
d) Claviceps purpurea
37. Match column I with column II and select the correct option

Column I (kingdom) Column II (Class)
A. Morels 1. Deuteromycetes
B. Smut
2. Ascomycetes
C. Bread mould
3. Basidiomycetes
D. Imperfect fungi
4. Pycomycetes
a) A-3, B - 4, C-1, D- 2
b) A-2, B- 3, C-4, D-1
c) A-3, B-4, C-2, D-1
d) A-2, B- 1, C-4, D-3
38. Which one single organism or the pair of organisms is correctly assigned to its or their named taxonomic group
a) Lichen is a composite organism formed from the symbiotic association of an algae and a protozoan
b) Yeast used in making bread and beer is a fungus
c) Nostoc and Anabaena are examples of Protista
d) Paramecium and Plasmodium belong to the same kingdom as that of penicillium
39. The imperfect fungi which are decomposers of litter and help in mineral cycling belong to
a) Deuteromycetes
b) Basidiomycetes
c) Phycomycetes
d) Ascomycetes
40. Choose the wrong statement
a) Pencillium is multicellular and Produces antibiotics
b)Neurospora is used in the study of biochemical genetics
c) Morels and truffles are poisonous mushrooms
d) Yeast is unicellular and useful in fermentation
41. Which one of the following is wrong for fungi
a) They are eukaryotic
b) All fungi possess a purely cellulosic cell wall
c) They are heterotrophic
d) They are both unicellular and multicellular
42. Which of the following organisms completely lack cellwall, they are the smallest living cells knownand can survive without oxygen
a) Mycoplasma
b) Euglenoids
c) Slime moulds
d) All of these
43. The type of diploid sexual resting spores formed during unfavorable conditions in Spirogyra is
a) Germ spores
b) Zygospores
c) Zoospores
d) Aplanospores
44. The living characteristic of virus is
a) Presence of nucleic acid as their genetic material
b) Ability to produce their own copies
c) Cellular organization
d) Autotrophic nutrition
45. Heterocyst present in Nostoc is specialized for
a) Fragmentation
b)Nitrogen fixation
c) Storage
d) Photosynthesis

## ZOOLOGY

46. Coordination between left and right cerebral hemispheres is brought out by
a)Corpus luteum
b)Corpus spongiosum
c) pons varolii
d)Corpus callosum
47. Human brain is considered evolutionarily advanced due to the development of
a)Diencephalon
b)Gyri and sulci
c)optic lobes
d)Medulla oblongata
48. Association areas of cerebral cortex are concerned with
a)Purely sensory functions
b)purely motor functions
c)Functions neither clearly sensory nor motor
d)Vision and equilibrium
49. Frontal, parietal, temporal and occipital lobes are the parts of
a)optic lobes
b)Cerebellum
c)thalamencephalon
d)Cerebrum
50. Anterior choroid plexus is present in
a)Roof of diencephalon
b)Roof of medulla
c)Floor of epithalamus
d)Floor of medulla
51. Centers for satiety, feeding and thirst are present in
a)medulla oblongata
b)hypothalamus
c) pons varolii
d)Cerebellum
52. Limbic system of brain does not include
a)Cerebral hemispheres
b)Amygdala
c)Arborvitae
d)Hippocampus
53. Dorsal portion of mesencephalon consists of
a)Posterior choroid plexus
b)Corpora quadrigemina
c)Arborvitae
d)Corpus callosum
54. Crus cerebrum connects
a)Paracoels with diocoel
b)Cerebellum with spinal cord
c)Cerebral hemispheres with pons
d)Cerebellar halves
55. 'The tree of life' is
a)Medulla oblongata
b)Arbor vitae
c)Pons varolii
d)Organ of corti
56. Pineal stalk arises from
a)Dorsal surface of epithalamus
b)Ventral surface of epithalamus
c) Ventral surface of hypothalamus
d)Dorsal surface f hypothalamus
57. Control and coordination of muscular movements is the function of
a)Mesencephalon
b)Hypothalamus
c)Cerebellum
d)Medulla
58. This one forms a bridge between two cerebellar hemispheres
a)Corpus callosum
b)Pons varolii
c)Cerebral peduncles
d)Infundibulum
59. Relay station between the cerebellum, spinal cord and rest of the brain is
a)Medulla oblongata
b)Mesencephalon
c)Cerebellum
d)Pons varolii
60. Brain stem does not include
a)Medulla oblongata
b)Mesencephalon
c) Hippocampus
d)Pons varolii
61. The vascular folded structure seen in the medulla oblongata is
a)Arbor vitae
b)Anterior choroid plexus
c)Posterior choroid plexus
d) Infundibulum
62. Iter lies in between
a)paracoel and diocoel b)metacoel and paracoel c)Paracoel and paracoel d)metacoel and diocoel
63. Neuronal processing related to language learning in right handed people is performed in this part of brain
a)left cerebral hemisphere
b)Right cerebral hemisphere
c)left cerebellar half
d)Right cerebellar half
64. Removal of which part of brain causes immediate death?
a)Cerebrum
b)Optic lobes
c)Medulla oblongata
d)Olfactory lobes
65. Study the following about the brain.
I. The midbrain, pons, cerebellum and medulla oblongata together form brain stem.
II. The inner parts of cerebral hemispheres, amygdala and hippocampus form the limbic system.
III. Each cerebral hemisphere is divided into frontal, parietal, temporal and occipital lobes.

Correct ones of the above are
a)Only I and III
b)Only I and II
c)Only II and III
d)I, II, III
66. Osmoregulatory and thermoregulatory centers are located in
a)Pons varolii
b)Hypothalamus
c)Medulla oblongata
d)Diencephalon
67. Match the following and choose the correct combination

## BRAIN STRUCTURE

A)Superior colliculi-hearing
B)Association areas
C)Inferior colliculi-auditing
D)Cerebellum

The correct answer is
a)A-3, B-2, C-4, D-5
b)A-3, B-2, C-4, D-1
c) A-2, B-4, C-1, D-5
d)A-3, B-2, C-1, D-5
68. Match the following and choose the correct combination.

CENTRE
A)Pnemotaxic centre
B) satiety centre
C)Vomiting centre
D)Arborvitae

The correct answer is
a)A-3, B-1, C-4, D-2
b)A-4, B-2, C-5, D-1
c) A-5, B-3, C-1, D-2
d)A-2, B-3, C-1, D-5
69. Photosensitive substance concerned with vision is
a)melanin
b)retinol
c)rhodpsin
d)collagen
70. In humans color vision is controlled by
a)rods
b)rhabdome
c)crystalline
d)cones
71. This part of eyes is useful much while walking,reading and driving
a)Optic disc
b)fovea
c)blind spot
d) sclera
72. Main function of vestibular apparatus is
a)Processing information of vision
c)Perceive the motion of head
73. Lens and cornea are nourished by
a) vitreous humor
b)Aqueous humor
c) sclera
d)canal of schlemm
74. Principle function of iris is
a)Filtering dust particles
b) Preventing entry of UV rays into eyes
c) Formation of tears
d)To regulate the amount of light that enter the eyes
75. Stapes is attached to
a)Fenestra ovalis
b)Eustachian tube
c)Fenestera rotunda
d)Tympanic membrane

## 76. Otolith organ comprises

a)Ampulla and crista
b)Cochlea and semicircular canals
c) sacculeand utricle
d)Ampulla and macula
77. Set of non-vascular parts of eyes is
a)Lens,cornea
b)Ciliarybody,iris
c)Choroid,cornea
d)Retina,lens
78. The correct pair with regard to human sense organs is
a)Optic disc-No rods but only cones
b)Macula-vestibule
c)Scotopic vision-Iodopsin
d)Tympanic cavity-Fluid filled cavity
79. Near or far vision in a normal person is attained by changing the
a)Position of retina
b)Area of pupil
c) Shape of lens
d)Position of cornea
80. Aqueous humor is secreted by
a)Lacrimal glands
b)Ciliary processes
c)Iris
d)Scleral venoussinus
81. Receptors for gravity are present on
a)Cochlea
b)Macula
c)Semicicular canals
d)Middle car
82. Organ of corti is located in
a)Sacculus
b)Scala tympani
c)Scala media
d)Scala vestibule
83. Eustachian tube connects
a) External ear and middle ear
b)Pharynx and middle ear
c)Middle ear and internal ear
d)Oral cavity and external ear
84. The correct order of arrangement of ear ossicles from outer to inner is
a)Malleus,stapes,incus
b)Incus,stapes,malleus
c)Stapes,malleus,incus
d)Malleus,incus,stapes
85. Aqueous and vitreous humors are separated by
a) Conjunctiva
b) Cornea
c) Iris
d) Lens
86. Eustachian tube helps in
a)Carry impulses to internal ear
b)Carry vibrations from external ear
c)Maintains body posture
d)Maintain equilibrium on either side of ear drum
87. Canal of schlemm is present at the junction of
a)Middle ear and pharynx
b)Sclera and cornea of eye
c)Lens and retina of eye
d) Fovea and optic disc
88. The layer closest to the choroid coat of eyes is
a)Photoreceptor layer
b)Bipolar cell layer
c) Ganglion cell layer
d)Multipolar cell layer
89. Visual purple is important in
a)Colour vision
b)Scotopic vision
c)Photopic vision
d)Binocular vision
90. The up-ward and down-ward movement in a lift are perceived by
a)Saccule
b)Utricle
c)Cochlea
d)Organ of corti

## PHYSICS

91. Work done in the given P-V diagram in the cyclic process is;
a) PV
b) 2 PV
c) $\mathrm{PV} / 2$
d) 3 PV

92. When a system is taken from state $i$ to a state $f$ along path iaf, $\mathrm{Q}=50 \mathrm{~J}$ and $\mathrm{W}=20 \mathrm{~J}$. Along path ibf, $Q=\mathbf{3 5} \mathbf{~ J}$. If $W=\mathbf{- 1 3} \mathbf{~ J}$ for the curved return path $f I$, $Q$ for this path is
a) 33 J
b) 23 J
c) -7 J
d) -43 J

93. When a monoatomic gas expands at constant pressure, the percentage of heat supplied that increases the internal energy of the gas and that which is involved in expansion is
a) $75 \%, 25 \%$
b) $25 \%, 75 \%$
c) $60 \%, 40 \%$
d) $40 \%, 60 \%$
94. In the following indicator diagram, the net amount of work done will be
a) Positive
b) Negative
c) Zero
d) Infinity

95. Efficiency of a Carnot engine is $\mathbf{5 0 \%}$ when temperature of outlet is 500 K . In order to increase efficiency up to $60 \%$ keeping temperature of intake the same what is temperature of outlet
a) 200 K
b) 400 K
c) 600 K
d) 800 K
96. If the door of refrigerator is kept open, then which of the following is true
a) Room is cooled
b) Room is heated
c) Room is either cooled or heated
d) Room is neither cooled nor heated
97. During an adiabatic compression, $\mathbf{8 3 0} \mathbf{J}$ of work is done on $\mathbf{2}$ moles of a diatomic ideal gas to reduce its volume by $\mathbf{5 0 \%}$. The change in its temperature is nearly $\left(R=8.3 \mathrm{JK}^{-1} \mathrm{~mol}^{-1}\right)$
a) 40 K
b) 33 K
c) 20 K
d) 14 K
98. The r.m.s. velocity of oxygen molecular at $16^{\circ} \mathrm{C}$ is $474 \mathrm{~m} / \mathrm{sec}$. The r.m.s velocity in $\mathrm{m} / \mathrm{s}$ of hydrogen molecule at $127^{\circ} \mathrm{C}$ is
a) 1603
b) 1896
c) 2230.59
d) 2730
99. At $0^{\circ} \mathrm{K}$ which of the following properties of a gas will be zero?
a) Kinetic energy
b) Potential energy
c) Vibrational energy
d) Density
100. The perfect gas equation for $\mathbf{4}$ gram of hydrogen gas is
a) $\mathrm{PV}=\mathrm{RT}$
b) $\mathrm{PV}=2 \mathrm{RT}$
c) $\mathrm{PV}=\frac{1}{2} \mathrm{RT}$
d) $P V=4 R T$
101. The number of degrees of freedom for each atom of a monoatomic gas is
a) 3
b) 5
c) 6
d) 1
102. In an adiabatic expansion of a gas initial and final temperatures are $T_{1}$ and $T_{2}$ respectively, then the change in internal energy of the gas is
a) $\frac{R}{\gamma-1}\left(T_{2}-T_{1}\right)$
b) $\frac{R}{\gamma-1}\left(T_{1}-T_{2}\right)$
c) $R\left(T_{1}-T_{2}\right)$
d)Zero
103. An ideal gas is taken around $A B C A$ as shown in the above $P-V$ diagram. The work done during a cycle is
a) 2 PV
b) PV
c) $1 / 2 P V$
d)Zero

104. The respective speeds of five molecules are $2,1.5,1.6,1.6$ and $1.2 \mathrm{~km} / \mathrm{sec}$. The most probable speed in $\mathrm{km} / \mathrm{sec}$ will be
a) 2
b) 1.58
c) 1.6
d) 1.31
105. In which of the following processes the internal energy of the system remains constant?
a) Adiabatic
b) isochoric
c) Isobaric
d) Isothermal
106. A given mass of a gas expands from the state $A$ to the state $B$ by three paths 1,2 and 3 as shown in the figure. If $W_{1}, W_{2}$ and $W_{3}$ respectively by the work done by the gas along the three paths then
a) $W_{1}>W_{2}>W_{3}$
b) $W_{1}<W_{2}<W_{3}$
c) $W_{1}=W_{2}=W_{3}$
d) $W_{1}<W_{2}=W_{3}$

107. On compressing a gas suddenly, its temperature
a) increases
b) decreases
c) remains constant
d) all the above
108. Two hollow spheres of same material one with double the radius of the other and double the thickness of the other filled with ice, The ratio of times in which ice gets melted in the two spheres is
a) $2: 1$
b) $1: 2$
c) $4: 1$
d) $1: 4$
109. Two spheres $A$ and $B$ with masses in the ratio $2: 3$ and specific heat $2: 3$ fall freely from rest. If the rise in their temperatures on reaching the ground are in the ratio $1: 2$ the ratio of their heights of fall is
a) $3: 1$
b) $1: 3$
c) $4: 3$
d) $3: 4$
110. A 50 kg man is running at a speed of $18 \mathrm{kmh}^{-1}$. If all the kinetic energy of the man can be used to increase the temperature of water from $20^{\circ} \mathrm{C}$ to $30^{\circ} \mathrm{C}$, how much water can be heated with this energy?
a) 15 g
b) 20 g
c) 30 g
d) 40 g
111. Find the external work done by the system in kcal, when 20 kcal of heat is supplied to the system and the increase in the internal energy is $8400 \mathrm{~J}(\mathrm{~J}=4200 \mathrm{~J} / \mathrm{kcal})$
a) 16 kcal
b) 18 kcal
c) 20 kcal
d) 19 kcal
112. Air expands from 5 litres to 10 litres at $\mathbf{2}$ atm pressure. External workdone is
a) 10 J
b) 1000 J
c) 3000 J
d) 300 J
113. The temperature of 5 moles of a gas at constant volume is changed from $100^{\circ} \mathrm{C}$ to $120^{\circ} \mathrm{C}$. The change in internal energy is 80 J . The total heat capacity of the gas at constant volume will be in joule/Kelvin is
a) 8
b) 4
c) 0.8
d) 0.4
114. A Steel ball of mass 0.1 kg falls freely from a height of 10 m and bounce to a height of 5.4 m from the ground. If the dissipated energy in this process is absorbed by the ball, the rise in its temperature is (specificheatofsteel $460 \mathrm{jkg}^{-1} \mathrm{k}^{-1}, g=19 \mathrm{~ms}^{-2}$ )
a) $0.01^{0} \mathrm{C}$
b) $0.1^{0} \mathrm{C}$
c) $1^{0} \mathrm{C}$
d) $1.1^{0} \mathrm{C}$
115. Cooking is difficult on mountains because
a) water boils at low temperature
b) water boils at high temperature
c) water does not boil
d) it is cool there
116. A large block of ice is placed on a table when the surroundings are at $0^{\circ} \mathrm{C}$
a)ice melts at the sides
b) ice melts at the top
c) ice melts at the bottom
d) ice does not melt at all
117. The ratio of densities of two substances is $2: 3$ and that of specific heats $1: 2$. The ratio of thermal capacities per unit volume is
a) $1: 2$
b) $2: 1$
c) $1: 3$
d) $3: 1$
118. Two liquids $A$ and $B$ of equal volumes have their specific heats in the ratio $2: 3$. If they have same thermal capacity, then the ratio of their densities is
a) $1: 1$
b) $2: 3$
c) $3: 2$
d) $5: 6$
119. Specific heat of aluminium is $0.25 \mathrm{cal} / \mathrm{g}-{ }^{0} \mathrm{c}$. The water equivalent of an aluminium is vessel of mass one kilogram is
a) $40 \mathrm{cal} /{ }^{0} \mathrm{c}$
b) $250 g$
c) $250 \mathrm{cal} /{ }^{0} \mathrm{c}$
d) 40 g
120. Two liquids $A$ and $B$ are at $30^{\circ} c$ and $20^{\circ} c$ respectively. When they are mixed in equal masses the temperature of the mixture is found to be $26^{\circ} c$. The ratio of specific heats is
a) $4: 3$
b) $3: 4$
c) $2: 3$
d) $3: 2$
121. If 10 g of the ice at $0^{\circ} c$ is mixed with 10 g of water at $100^{\circ} c$, then the final temperature of the mixture will be
a) $5^{0} \mathrm{c}$
b) $10^{\circ} \mathrm{c}$
c) 100 K
d) $0^{0} c$
122. The heat energy required to vapourise 5 kg of water at 373 K is
a) $2700 \mathrm{~K} . \mathrm{cal}$
b) 1000 K.cal
c) $27 \mathrm{~K} . \mathrm{cal}$
d) 270 K.cal
123. A metal block absorbs 4500 cal of heat when heated from $30^{\circ} c$ to $80^{\circ} c$. Its thermal capacity is
a) 90 gm
b) $90 \mathrm{cal} /{ }^{0} \mathrm{C}$
c) 9 gm
d)) $9 \mathrm{cal} /{ }^{0} \mathrm{C}$
124. Three liquids $A, B$ and $C$ of masses $400 \mathrm{gm}, 600 \mathrm{gm}$ and 800 gm are at $30^{\circ} \mathrm{c}, 40^{\circ} \mathrm{c}$ and $50^{\circ} \mathrm{c}$ respectively. When $A$ and $B$ are mixed resultant temperature is $36^{\circ} c$ are mixed resultant temperature is $44^{\circ} c$ Then ratio of their specific heats are
a) $2: 1: 1$
b: $3: 2: 1$
c) $2: 2: 1$
d) $1: 4: 9$
125. 50 g of copper is heated to increase its temperature by $10^{\circ} \mathrm{C}$. If the same quantity of heat is given to 10 g of water, the rise in its temperature is
$\left(S_{c u}=420 \mathrm{~J} / \mathrm{kg} /{ }^{0} \mathrm{C}\right.$ and $S_{w}=4200 \mathrm{~J} / \mathrm{kg} /{ }^{0} \mathrm{C}$ )
a) $5^{0} \mathrm{C}$
b) $6^{\circ} \mathrm{C}$
c) $7^{0} \mathrm{C}$
d) $8^{0} \mathrm{C}$
126. Power of man who can chew 0.3 kg ice in one minute is (in cal/s)
a) 400
b) 4
3) 24
d) 240
127. One mole of oxygen is heated at constant pressure starting at $0^{\circ} \mathrm{C}$. The heat energy that must be supplied to the gas to double its volume ( $R$ is the molar gas constant)
a) $2.5 \times 273 \times R$
b) $3.5 \times 273 \times R$
c) $2.5 \times 546 \times R$
d) $3.5 \times 546 \times R$
128. A carnot's engine whose sink is at a temperature of 300 K has an efficiency $\mathbf{o f} \mathbf{4 0 \%}$. By how much should the temperature of the source be increased so as to increase the efficiency to $\mathbf{6 0 \%}$
a) 250 K
b) 275 K
c) 300 K
d) 325 K
129. An ideal Carnot's engine whose efficiency is $\mathbf{4 0 \%}$ receives heat at 500 K . If the efficiency to be $50 \%$ then the temperature of sink will be
a) 600 K
b) 800 K
c) 1000 K
d) 250 K
130. The average kinetic energy of a molecule of a gas at absolute temperature $T$ is proportional to
a) $1 / T$
b) $\sqrt{T}$
c) $T$
d) $T^{2}$
131. At a given temperature if $V_{r m s}$ is the root mean square velocity of the molecules of a gas and $V_{S}$ be the velocity of sound in it, then these are related as $\left(\gamma=\frac{C_{P}}{C_{V}}\right)$
a) $v_{r m s}=v_{S}$
b) $v_{r m s}=\sqrt{\frac{3}{\gamma}} \times v_{S}$
c) $v_{r m s}=\sqrt{\frac{\gamma}{3}} \times v_{S}$
d) $v_{r m s}=\sqrt{\frac{3}{\gamma}} \times v_{S}$
132. The average degrees of freedom per molecule for a gas is 6 . The gas performs 25 J of work when it expands at constant pressure. The heat absorbed by gas is
a) 75 J
b) 100 J
c) 150 J
d) 125 J
133. The root mean square velocity, $v_{r m s}$ the average velocity $v_{a v}$ and the most probable velocity, $v_{m p}$ of the molecules of the gas are in the order :
a) $v_{m p}>v_{a v g}>v_{r m s}$
b) $v_{r m s}>v_{\text {avg }}>v_{m p}$
c) $v_{\text {avg }}>v_{m p}>v_{r m s}$
d) $v_{m p}>v_{r m s}>v_{a v g}$
134. If number of molecules of $H_{2}$ are double than that of $\mathrm{O}_{2}$, then ratio of kinetic energy of hydrogen and that of oxygen at 300 K is
a) $1: 1$
b) $1: 2$
c) $2: 1$
d) $1: 16$
135. Air expands from 5 litres to 10 litres at 2 atm pressure. External workdone is
a) 10 J
b) 1000 J
c) 3000 J
d) 300 J

## CHEMISTRY

136. 0.45 gm of an organic compound containing only carbon, hydrogen and nitrogen on combustion gave 1.1 g of $\mathrm{CO}_{2}$ and 0.3 g of water, the empirical formula of the compound?
a) $\mathrm{C}_{4} \mathrm{H}_{3} \mathrm{~N}_{2}$
b) $\mathrm{C}_{3} \mathrm{H}_{4} \mathrm{~N}_{2}$
c) $\mathrm{C}_{4} \mathrm{H}_{3} \mathrm{~N}$
d) $\mathrm{C}_{3} \mathrm{H}_{4} \mathrm{~N}$
137. 


a) E-isomer
b) Z-isomer
c) Cis-isomer
d) Trans-isomer
138. The number of $\sigma$ and $\pi$ bonds in a molecule of acetonitrile are respectively
a) 2,5
b) 3,4
c) 4,3
d) 5,2
139. The optical inactivity due to internal compensation can be exhibited by:
a) dextro-tartaric acid
b) laevo-tartaric acid
c) racemic-tartaric acid d) meso-tartaric acid
140. The structure of allylchoride is
a) $\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2} \mathrm{Cl}$
b) $\mathrm{CH}_{2}=\mathrm{C} \mathrm{Cl}-\mathrm{CH}_{3}$
c) $\mathrm{CH} \mathrm{Cl}=\mathrm{CH}-\mathrm{CH}_{3}$
d) $\mathrm{CH} \mathrm{Cl}=\mathrm{C} \mathrm{Cl}-\mathrm{CH}_{3}$
141. Which of the following is hetero cyclic compound?
a) Anthracene
b) Pyrrole
c) Phenol
d) Isobutylene
142. Distillation is used to separate liquids which differ in the b.p by
a) $5^{0} \mathrm{C}$
b) $10^{\circ} \mathrm{C}$
c) $30-80^{\circ} \mathrm{C}$
d) $100^{\circ} \mathrm{C}$
143. Example of chain isomerism are
a) Ortho and meta toluic acids
b) Methyl acetate and ethyl formate
c) Pentanoic acid and 2-methyl butanoic acid
d) 2-pentanone and 3-pentanone
144. How many stereoisomers does this molecule have $\mathrm{CH}_{3} \mathrm{CH}=\mathrm{CHCH}_{2} \mathrm{CHBrCH}_{3}$
a) 4
b) 6
c) 8
d) 2
145. In the hydrocarbon

$$
\begin{array}{cccccc}
\mathrm{CH}_{3}-\mathrm{CH}=\mathrm{CH}-\mathrm{CH}_{2}-\mathrm{C} \equiv \mathrm{CH} \\
6 & 5 & 4 & 3 & 2 & 1
\end{array}
$$

The state of hybrization of carbons $\mathbf{1 , 3}$ and 5 are in the following sequence:
a) $s p^{2}, s p, s p^{3}$
b) $s p, s p^{3}, s p^{2}$
c) $s p, s p^{2}, s p^{3}$
d) $s p^{3}, s p^{2}, s p$
146. Which of the following compounds will exhibit cis-trans (geometrical) isomerism?
a) Butanol
b) 2-Butyne
c) 2-Butenol
d) 2-Butene
147. The correct decreasing order of priority for the functional groups of organic compounds in the IUPAC system of nomenclature is
a) $-\mathrm{COOH},-\mathrm{SO}_{3} \mathrm{H}-\mathrm{CONH}_{2},-\mathrm{CHO}$
b) $-\mathrm{SO}_{3} \mathrm{H},-\mathrm{COOH},-\mathrm{CONH}_{2},-\mathrm{CHO}$
c) $-\mathrm{CHO},-\mathrm{COOH},-\mathrm{SO}_{3} \mathrm{H},-\mathrm{CONH}_{2}$
d) $-\mathrm{CONH}_{2},-\mathrm{CHO},-\mathrm{SO}_{3} \mathrm{H},-\mathrm{COOH}$
148. The IUPAC name of the following compound is
a) 4-Bromo-3-cyanophenol
b) 2-Bromo-5-hydroxybenzonitrile
c) 2-Cyano-4-hydroxybromobenzene
d) 6-Bromo-3-hydroxybenzonirile

149. In Duma's method for estimation of nitrogen, 0.25 g of an organic compound gave 40 mL of nitrogen collected at 300 K temperature and 725 mm pressure. If the aqueous tension at 300 K is $\mathbf{2 5} \mathbf{~ m m}$, the percentage of nitrogen in the compound is:
a) 18.20
b) 16.76
c) 15.76
d) 17.36
150. The number of isomers of the compound $\mathrm{C}_{2} \mathrm{BrFCl}$ is
a) 3
b) 4
c) 5
d) 6
151. Correct order of stability is:
a) cis-2-butene.1-butene>trans-2-butene
b) trans-2-butene>cis-2-butene>1-butene
c) 1-butene $>$ cis-2-butene $>$ trans-2-butene
d) cis-2-butene $>$ trans-2-butene $>1$-butene
152. Which of the following does not exhibit cis-trans isomerism
a) $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}=\mathrm{NOH}$
b) $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{~N}=\mathrm{NC}_{6} \mathrm{H}_{5}$
c) $><$
d) None of these
153. 0.37 g of a given compound gave 0.631 g of the silver bromide, the percentage of bromine in it
a) $52.3 \%$
b) $72.6 \%$
c) $80.2 \%$
d) $28.4 \%$
154. During hearing of a court case, the judge suspected that some changes in the documents had been carried out. He asked the forensic department to check the ink used at two different places.
According to you which technique can give the best results?
a) Column chromatography
b) Solvent extraction
c) Distillation
d) Thin layer chromatography
155. The principle involved in paper chromatography is
a) adsorption
b) partition
c) solubility
d) volatility
156. What is the hybridisation state of benzyl carbonium ion

a) $s p^{3}$
b) $s p^{2}$
c) $s p d^{2}$
d) $s p^{2} d$
157. Nitrogen detection in an organic compound is carried out by Lassaigne's test. The blue colour formed corresponds to which of the following formulae
a) $\mathrm{Fe}_{3}\left[\mathrm{Fe} \mathrm{CN}_{6}\right]_{3}$
b) $\mathrm{Fe}_{3}\left[\mathrm{Fe} \mathrm{CN}_{6}\right]_{2}$
c) $\mathrm{Fe}_{4}\left[\mathrm{Fe} \mathrm{CN}_{6}\right]_{3}$
d) $\mathrm{Fe}_{4}\left[\mathrm{Fe} \mathrm{CN}_{6}\right]_{2}$
158. In the Kjeldahl's method for estimation of nitrogen present in a soil sample, ammonia evolved from 0.75 g of sample neutralized 10 mL of $1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$. The percentage of nitrogen in the soil is
a) 37.33
b) 45.33
c) 35.33
d) 43.33
159. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}$ is a functional isomers of
a) $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OCH}_{3}$
b) $\mathrm{CH}_{3} \mathrm{OC}_{3} \mathrm{H}_{7}$
c) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}$
d) $\mathrm{CH}_{3} \mathrm{CHOHCH}_{3}$
160. The IUPAC name of $\mathrm{CH}_{3}-\mathrm{C} \equiv \mathrm{C}-\mathrm{CHCH} 3_{2}$ is
a) 4-methyl-2-pentyne
b) 4,4-dimethyl-2-butyne
c) methyl isopropyl acetylene
d) 2-methyl-4-pentyne
161. Assertion (A): Lactic acid is optically active compound

Reason ( $R$ ): It contains a chiral centre with plane of symmetry
a) Both $A$ and $R$ are true and $R$ is correct explanation of $A$
b) Both $A$ and $R$ are true and $R$ is not correct explanation of $A$
c) $A$ is true but $R$ is false
d) $A$ is false but $R$ is true
162. The geometrical isomerism is shown by
a)

b)

c)

d)

163. Which of the following compounds contains $1^{0}, 2^{0}, 3^{0}$ as well as $4^{0}$ carbon atoms
a) Neopentane
b) 2-methyl pentane
c) 2,3-dimethyl butane
d) 2,2,3-trimethyl pentane
164. Correct configuration of the following as
a) $3 \mathrm{~S}, 2 \mathrm{~S}$
b) $3 \mathrm{~S}, 2 \mathrm{R}$
c) $3 R, 2 R$
d) $3 R, 2 S$

165. The correct name for the following hydrocarbon is
a) Tricyclo $[4,1,0]$ heptane
b) Bicyclo $[5,2,1]$ heptane
c) Bicyclo $[4,1,0]$ heptane
d) Bicyclo $[4,1,0]$ hexane

166. The Beilstein test for organic compounds is used to detect
a) nitrogen
b) sulphur
c) carbon
d) halogens
167. The process of separation of a racemic modification into d and $l$-enantiomers is called
a) Resolution
b) Dehydration
c) Revolution
d) Dehydrohalogenation
168. Identify the compound that exhibits tautomerism
a) 2-pentanol
b) Phenol
c) 2-butanone
d) Lactic acid
169. Lassaigne's test for the detection of nitrogen fails in
a) $\mathrm{NH}_{2} \mathrm{CONHNH}$
b) $\mathrm{NH}_{2} \mathrm{NH}_{2} \cdot \mathrm{HCl}$
c) $\mathrm{NH}_{2} \mathrm{CONH}_{2}$
d) $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{NHNH}_{2} \cdot \mathrm{HCl}$
170. The chirality of the compound
a) $R$
b) S
c) E
d) Z

171. Which of the following is optically inactive

b)

c)

d) None of these
a)
172. The most suitable method for separation of a $1: 1$ mixture of ortho and para nitrophenols is
a) sublimation
b) chromatography
c) crystallization
d) steam distillation
173. Which of the following is optically active

a)
b) $\mathrm{CH}_{3}-\mathrm{CHOH}-\mathrm{CH}_{3}$

d)
$\mathrm{CH}_{3}-\mathrm{CH}_{2}-\underset{\substack{\mathrm{C} \\ \mathrm{OH} \\ \mathrm{OH}}}{\mathrm{C}}=\mathrm{O}$
174. The correct IUPAC name for
a)5-methyl-4- 1,2-dimethylpropyl heptane
b) 3-methyl-4- 1,2-dimethylpropyl heptane
c) 2,3,5-trimethyl-4-prophylheptane

d) 4-propyal-2,3,5-trimethylpeptane
175. Total number of stereo isomers (optical and geometrical) of the following compound are $\mathrm{CHCl}=\mathrm{CH}-\mathrm{CH} \mathrm{CH}_{3} \mathrm{C}_{2} \mathrm{H}_{5}$
a) 2
b) 4
c) 6
d) 8
176. The number of ether isomers possible for $\mathrm{C}_{4} \mathrm{H}_{10} \mathrm{O}$ are
a) 2
b) 5
c) 4
d) 3
177. Keto-enol tautomerism is observed in:
a) $\mathrm{C}_{5} \mathrm{H}_{6}-\stackrel{O}{\|}-\mathrm{CH}_{2}-\stackrel{O}{\|}-\mathrm{C}_{6} \mathrm{H}_{5}$
b) $\mathrm{H}_{5} \mathrm{C}_{6}-\stackrel{\stackrel{O}{\mathrm{C}}-\mathrm{CH}_{3}}{ }$
c) $H_{5} \mathrm{C}_{6}-\stackrel{\stackrel{O}{\|}}{\mathrm{C}}-\mathrm{H}$
d) Both (a) and (b)
178. The best method for the separation of naphthalene and benzoic acid from their mixture is
a) distillation
b) sublimation
c) chromatography
d) crystallization
179. If a compound has $n$ asymmetric carbon atoms, then maximum number of optical isomers are given by the formula
a) $2^{n}$
b) $\left(\frac{1}{2}\right)^{n}$
c) $\sqrt{2 n}$
d) $2 \sqrt{n}$
180. The correct statement about the compound $(\mathrm{A}),(\mathrm{B})$ and $(\mathrm{C})$ is:

(A)

(B)
a) (A) and (B) are identical
c) (A) and (C) are enantiomers

(C)
b) (B) and (C) are diastereomers
d) (A) and (B) are enantiomers

## ALL THE BEST

Adyar, Chennai is an initiative that was started to bring the quality of the most experienced faculty team in Tamil Nadu to your own NEET preparation process. We are releasing a chapter wise test series every week. Download them and take the test! When the answer key is released the next day, you get to evaluate yourself.

| Test | Date | BOTANY | ZOOLOGY | PHYSICS | CHEMISTRY |
| :---: | :---: | :---: | :---: | :---: | :---: |
| TEST-1 | 2/9/2018 | Morphology of Flowering Plants ( Root, Stem, Leaf, Flower, Inflorescence) | The Living World | Physical World, Units and Measurements, | Structure of Atom |
| TEST-3 | 16/09/2018 | Morphology of Flowering Plants (Fruits \& Taxonomy) | Animal Kingdom Non Chordates | Motion in a Straight Line | Classification of Elements and Periodicity in Properties |
| TEST - 5 | 30/09/2018 | Anatomy of Flowering Plants | Animal Kingdom Chordates - Fishes, Amphibia, Reptilia, Aves \& Mammalia | Motion in a Plane | Chemical Bonding and Molecular Structure |
| TEST - 7 | 14/10/2018 | Cell: The Unit of Life | Animal Kingdom - <br> Earth Worm, Cockroach, Frog | Laws of Motion | States of Matter |
| TEST-9 | 28/10/2018 | Biomolecules, Cell Cycle and Cell Division | Structural Organisation | Work, Energy and Power | Thermodynamics |
| TEST-11 | 11/11/2018 | Plant PhysiologyTransport in Plants | Human Physiology: Digestion and Absorption | System of Particles and Rotational Motion | Chemical Equilibrium and Ionic Equilibrium |
| TEST - 13 | 25/11/2018 | Mineral Nutrition | Breathing and Exchange of Gases | Gravitation | Some Basic Concepts of Chemistry, Redox Reactions |
| TEST - 15 | 9/12/2018 | Photosynthesis in Higher Plants | Body Fluids and Circulation | Mechanical Properties of Solids | Hydrogen \& Its Compounds |
| TEST-17 | 23/12/2018 | Respiration in Plants | Elimination of Excretory Products | Mechanical Properties of Fluids | The S-Block Elements (IA \& IIA Group Elements) |
| TEST - 19 | 6/1/2019 | Plant Growth and Development | Locomotion and Movement | Thermal Properties of Matter | The P-Block Elements (IIIA \& IVA Group Elements) |
| TEST - 21 | 20/1/2019 | Biological Classification | Neural Control and Co-ordination | Thermodynamics, Kinetic Theory, | Organic Chemistry Some Basic Principles and Techniques |
| TEST-23 | 3/2/2019 | Plant Kingdom | Chemical Co-ordination and Integration | Oscillations, Waves | Hydrocarbons Alkanes, Alkenes, Alkynes \& Benzene |


| Test | Date | BOTANY | ZOOLOGY | PHYSICS | CHEMISTRY |
| :---: | :---: | :---: | :---: | :---: | :---: |
| TEST - 2 | 9/9/2018 | Reproduction in Organisms - Plants | Human Reproductive System | Electric Charges and Fields, | The Solid State |
| TEST-4 | 23/9/2018 | Sexual Reproduction in Flowering Plants | Reproductive Health | Electrostatic Potential and Capacitance | Dilute Solutions |
| TEST-6 | 7/10/2018 | Sexual Reproduction in Flowering Plants | Principles of Inheritance and Variation | Current Electricity | Electrochemistry |
| TEST - 8 | 21/10/2018 | Principles of Inheritance and Variation | Molecular Basis of Inheritance | Moving Charges and Magnetism | Chemical Kinetics |
| TEST - 10 | 4/11/2018 | Principles of Inheritance and Variation | Evolution | Magnetism and Matter | Surface Chemistry |
| TEST - 12 | 18/11/2018 | Molecular Basis of Inheritance | Human Health and Diseases | Electromagnetic Induction, Alternating Current | General Principles and Processes of Isolation of Elements |
| TEST - 14 | 2/12/2018 | Molecular Basis of Inheritance | Strategies for Enhancement in Food Production | Electromagnetic Waves, Communication Systems | The P-Block Elements |
| TEST - 16 | 16/12/2018 | Strategies of Enhancement in Food Production | Biotechnology and its Applications | Ray Optics and Optical Instruments | The D-And F-Block Elements, Coordination Compounds |
| TEST - 18 | 30/12/2018 | Microbes in Human Welfare | Organisms and Population | Wave Optics | Haloalkanes and Arenes, Alcohols, Phenols and Ethers |
| TEST - 20 | 13/1/2019 | Biotechnology - <br> Principles and Processes | Eco System | Dual Nature of Radiation and Matter | Aldehydes, Ketones and Carboxylic Acids |
| TEST - 22 | 27/1/2019 | Biotechnological Applications in Medicine | Biodiversity and Conservation | Atoms, Nuclei | Amines, Chemistry in Everyday Life |
| TEST - 24 | 10/2/2019 | Plant Ecology | Environmental Issues | Semiconductor Electronics: Materials, Devices and Simple Circuits | Biomolecules, Polymers |




Admission No. $\square$ Section- $\square$

Name $\square$
$\square$
Signature of the Candidate Instructions:

1. Use Blue / Black ball points pen only for writing particulars/marking.
2. Darken the Circle in the space provided only.
3. Use of White fluid or any other material, which damages the answer sheet, is not allowed.
4. Do not staple, tear or scribble on the answer sheet.

(1) (2)


Test Centre

Correct Method
(A)

Wrong Method
(x) (C) (D)

