20th Jan, 2019





SYLLABUS



IMPORTANT INSTRUCTIONS:

	1	There are four parts	IMPORTANT IN	<u>STRUCTIONS:</u> B C and D consistin	g of Botany Zoology Phys	sics and			
	1.	Chemistry having 45 questions in each part of equal weightage. Each question is allotted 4 (four)							
		marks for each cor	rect response.		· Zuen question is allowed	(1001)			
	2.	Candidates will be awarded marks as stated above in instruction for correct response of each							
		1/4 (one fourth) m	arks will be deducted	for indicating incorr	ect response of each quest	ion. No			
		deduction from the t	otal score will be made in	f no response is indicate	ed for an item in the answer s	heet.			
	3.	There is only one of	correct response for each	h question. Filling up	more than one response	in each			
		question will be t	treated as wrong resp	oonse and marks for	wrong response will be d	leducted			
	accordingly as per instruction.								
			BOTA	ANY					
	1.	During Gram's stat	in Z						
		a) All bacteria wheth	her Gram (+) ve or (-ve),	take crystal violet stain	1				
		b) Only gram +ve ba	icteria take crystal violet	stain					
		c) Only Gram – ve b	acteria take crystal viole	t stain	1				
		d) Gram (+) ve bac	terna lose this stain after a	alconol treatment and ta	ake red stain of safranin				
	2	Under ontimum co	ndition Bacterial cells (livide once in 20 min	ites by hingry fission How	many			
	4.	Under optimum condition Bacterial cells divide once in 20 minutes by binary fission. How many bacteria will be produced in 2 hours with some rate of division 2							
		a) 8	b) 32	c) 128	d) 64				
			0,02	•) 120					
g.com	3.	The archaebacteria occurring in marshes, swamps, rumens of cattles, gobar gas plants are							
		a) methanogens	b) ammonifying bact	eria c) thermoacidor	bhiles d) denitrifying bacteria	ı			
	4.	A symbiotic nitroge	en fixing moneran amoi	ng the following is					
		a) Nitrocystis	b) Anabaena	c) Nitrobacter	d) Escherichia				
Ĩ									
pi	5.	From where you wi	ill collect E. coli ?						
		a) Human excreta	b) On leaves	c) Water	d) Human stomach				
08	6	A bacterial call divides every minute. It takes one hour to fill a sun. Time required to fill half our is							
	0.	a) 60 minutes	b) 59 minutes	c) 30 minutes	d) 29 minutes	up is			
		a) of minutes	b) 57 minutes	c) 50 minutes	d) 2) minutes				
	7.	The joker of plant l	kingdom area						
		a) Bacteria	b) Archaebacteria	c) PPLO	d) Viriods				
		,	,		,				
	8.	Which one of the fo	Which one of the following statements is wrong						
		a) Golden algae are a	also called desmids	b) Eubacteria are al	so called false bacteria				
		c) Phycomycetes are	e fungi	d) Cyanobacteria an	re also called blue – green ala	ıge			
	9.	Select the wrong sta	atement						
		a) Bacterial cell wall	I is made up of peptidogly	ycan	_				
		b) Pili and fimbriae	are mainly involved in m	othity of bacterial cells	8				
		d) Mycoplasma is a	k flageflated cells	n					
		u) wrycopiasina is a	wan – iess interoorganisi	11					
	10.	Which one of the fo	ollowing is a slime moul	d					
	-	a) Thiobacillus	b) Anabaena	c) Rhizopus	d) Physarum				
				· •					
	11.	The 'fire' algae res	ponsible for red tides a	e the red dinoflagella	tes, which are				
1		a) Ceratium	b) Gonyaulax	c) Gymnodinium	d) (2) and (3)				

	a) Dinoflagellates	b) Diatoms	c) Zooflagellates	d) Sporozoans		
13.	Paralytic shellfish po	oisoning (PSP) is	caused by toxin saxitonin	by		
	a) Vorticella	b) Ephidicum	c) Gonyaulax	d) Ceratium		
14	Which animalcule is	immortal ?				
14.	a) Paramecium	b) Plasmodiun	n c) Amoeba	d) Euglena		
	u) i ululloolulli	<i>c)</i> i iusino uiun	c) / IIII000u	d) Eugleria		
15.	Match the following	and select the co	orrect combination from th	ne options given below		
	Column	I	Column II (Class)			
	(kinge	lom)				
	A. Planta	.e 1. A	Archaebacteria			
	B. Fungi	2. I	Euglenoids			
	C. Protist	ta 3. I	Phycomycetes			
	D. Money	ra 4. A	Algae			
	a) A-4, B – 3, C- 2, D	- 1 b) A- 1, B- 2,	C-3, D-4 c) A-3, B-4, C-2,	, D-1 d) A-2, B-3, C-4, D-		
16	Which of the followi	ng doos not bolo	ng to the kingdom Prostie			
10.	a) Chrysophytes	b) Fuglenoids	c) Ascomycetes	d) Dinoflagellates		
	a) Chrysophytes	b) Euglenoids	c) Ascomycetes	d) Dinonagenates		
17.	In the five kingdom	classification, ch	lamydomonas and chlorel	lla have been included in		
	a) Protista	b) algae	c) Plantae	d) monera		
18.	The beautiful diatom	ıs and desmids a	re placed under			
	a) chrysophytes	b) dinoflagella	tes c) euglenoids	d) slime moulds		
10						
19.	In which group of or	ganisms the cell	walls form two thin overl	apping shells which fit toge		
	a) Chrysophytes	b) Euglenoids	c) Dinoflagellates	a) Slime moulds		
20	Chrysonhytes Fugle	onoids Dinoflage	llates and Slime moulds a	re included in the kingdom		
20.	a) Protista	h) Fungi	c) Animalia	d) Monera		
	a) i fotista	b) I uligi	c) / Ammana	d) Wollera		
21.	Select the wrong statement					
	a) The walls or diator	ms are easily dest	ructible			
	b) 'Diatomaceous earth' is formed by the cell walls of diatoms					
	c) Diatoms are chief r	producers in the o	ceans			
	d) Diatoms are microscopic and float passively in water					
22.	Mycorrhiza is					
	a) symbiotic association	ion of a soil fung	us and roots of higher plant	ts		
	b) Parasite associatior	ı between a fungu	is and roots of seeded plants	3		
	c) saprophytic associa	tion between a fu	ingus and roots of seeded pla	ants		
	d) symbiotic association between an algae and fungi					
				y known og		
23.	In sac fungi (ascomv	cetes) the ascosn	ores occur in sac like bod	V KHOWH AS		
23.	In sac fungi (ascomy a) ascus	cetes) the ascosp b) basidium	c) ascocarp	d) basidiocarp		
23.	In sac fungi (ascomy a) ascus	cetes) the ascosp b) basidium	c) ascocarp	d) basidiocarp		
23. 24.	In sac fungi (ascomy a) ascus The name Club fung	cetes) the ascosp b) basidium i is given to basi	c) ascocarp	d) basidiocarp		
23. 24.	In sac fungi (ascomy a) ascus The name Club fung a) Club shaped basidia	cetes) the ascosp b) basidium i is given to basi	diomycetes due to the pres b) club shaped basidios	d) basidiocarp sence of		

25.	All members of fungi imperfecti (Deutero-mycetes) group lack						
	a) sexual reproducti	on b) spores c) a	asexual repro	oduction	d) hyphae		
26.	Rice crop was destroyed by a fungus which resulted in Bengal famine (1942-43). It was due to a) Xanthomonas malvacearum b) Pyricularia oryzae c) Helminthosporium oryzea d) Puccinia graminis						
27.	A fungus whose ex drug (LSD) is	xtract of sclerotia can be c	chemically a	ltered to pr	oduce powerful hallucinogenic		
	a) Claviceps purpur	ea b) Aspergillus flavus	c) Amanit	ta caesarea	d) Psilocybe mexicana		
28.	Meiosis occurs in t	he life cycle of Rhizopus	during				
	a) formation of gem	etangium	b) germin	ation of zygo	ospore		
	c) formation of gerr	n sporangium	d) formation of aplanospores				
29.	Ascus in ascomyce a) two meiosis b)	tes contains 8 ascospores. one meiosis and one mitosi	These are f is c) four mit	Cormed as a tosis d) one	result of meiosis and two mitosis		
30.	Champ connection	s occur in					
	a) Haplomycetes	b) Saccharomycetes	c) Basidic	omycetes	d) Ascomycetes		
31.	1 The fruiting body formed from a filementous betweetrenbig organism which is known for its						
011	nutritive value for	the humanity is		pine of guing			
	a) Cremocarp	b) Acervulus	c) Basidic	ocarp	d) Akinete		
	u) crono curp		•) 2 451410	P			
32.	Find the correct m	atch					
	Name	Feature		Class			
	(a) Aspergillus	Aseptate mycelium	1	Ascomycetes	5		
	(b) Trichoderma	Imperfect fungi]	Phycomycete	es		

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33. Match the columns

(c) Rhizopus

(d) Puccinia

(A)	Early Blight of potato	(i)	Puccinia graminis
(B)	Late Blight to potato	(ii)	Ustilago tritici
(C)	Smut of Wheat	(iii)	Phytophthora infestans
(D)	Rust of Wheat	(iv)	Alternaria solani

Branched and septate mycelium

Coenocytic mycelium

a) (A)-(i), (B) – (ii), (C) – (iii), (D) – (iv) c) (A) - (ii), (B) - (iii), (C) - (iv), (D) - (i) b) (A) - (iv), (B) - (iii), (C) - (ii), (D) - (i)d) (A) - (iv), (B) - (ii), (C) - (iii), (D) - (i)

Deuteromycetes

Basidiomycetes

34. Find the correct match

a) Phytophthora infestans - Early blight of potato b) Albugo candida - white rust of crucifers c) Puccinia graminis - Loose smut of wheat d) All of the above

35. Which one of the following is endogenously produced

a) Ascospores b) Basidiospores c) Conidiospores d) All of the above

36. The staple crop in Ireland was completely destroyed in 1844 – 1846, by a fungal disease resulting in a great famine. The causal organism in this case was a) Puccinia graminis b) Ustilago tritici

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c) Phytophthora infestans d) Claviceps purpurea

	37.	Match column I with	n column II and select	the correct option			
		Column I (kingdom)	Column II (Class)	T			
		A. Morels	1. Deuteromycetes				
		B. Smut	2. Ascomvcetes				
		C. Bread mould	3. Basidiomycetes				
		D. Imperfect fungi	4. Pycomycetes				
		a) A-3, B - 4, C- 1, D-	2 b) A- 2, B- 3, C-4, D	D-1 c) A-3, B-4, C-2,	D-1 d) A-2, B- 1, C- 4, D-3		
			,	,			
	38.	Which one single org	anism or the pair of o	rganisms is correctly a	assigned to its or their named		
		taxonomic group					
		a) Lichen is a composi-	te organism formed from	m the symbiotic associa	ation of an algae and a protozoan		
		b) Yeast used in maki	ng bread and beer is a fu	ungus			
		c) Nostoc and Anabae	na are examples of Proti	ista			
		d) Paramecium and Pl	asmodium belong to the	same kingdom as that	of penicillium		
	39.	The imperfect fungi	which are decomposer	s of litter and help in r	nineral cycling belong to		
	071	a) Deuteromycetes	b) Basidiomycetes	c) Phycomycetes	d) Ascomycetes		
	40.	Choose the wrong st	atement				
		a) Pencillium is multicellular and Produces antibiotics					
		b)Neurospora is used in the study of biochemical genetics					
		c) Morels and truffles are poisonous mushrooms					
E		d) Yeast is unicellular	and useful in fermentati	ion			
U	41	Which one of the following is wrong for fungi					
5	41.	Which one of the foll	owing is wrong for fun				
b 0		a) They are eukaryotic		b) All fungi possess	a purely cellulosic cell wall		
E.		c) They are heterotrop	hic	d) They are both uni	cellular and multicellular		
	42.	Which of the following	ng organisms complete	lv lack cellwall, they a	re the smallest living cells		
ac		knownand can surviv	ve without oxygen		8		
		a) Mycoplasma	b) Euglenoids	c) Slime moulds	d) All of these		
tc			ý U	,	,		
$\mathbf{\lambda}$	43.	The type of diploid sexual resting spores formed during unfavorable conditions in Spirogyra is					
		a) Germ spores	b) Zygospores	c) Zoospores	d) Aplanospores		
	4.4	The Bailton a channe a familie	-4'6				
	44.	The living characteri	suc of virus is	tarial h) Ability to mus	duce their own entite		
		a) resence of nucleic actuals their genetic material b) Ability to produce their own copies					
		c) Cellular organization d) Autotrophic nutrition					
	45.	Heterocyst present ir	Nostoc is specialized f	for			
		a) Fragmentation	b)Nitrogen fixation	c) Storage	d) Photosynthesis		
	ZOOLOGY						
	46.	Coordination betwee	n left and right cerebra	al hemispheres is brou	ight out by		
		a)Corpus luteum	b)Corpus spongiosum	n c)pons varolii	d)Corpus callosum		
	17	Uumon huoin is ac	idanad avalutionaril	duonood duo 40 4ko J-	valormont of		
	4/.	a)Diencenhalon	b)Gyri and sulei	c)optic lobes	d)Medulla oblongata		
		a) 2 rencephaton	sy cyrr und bulor	0,0000	announna obrongum		
4	48.	Association areas of	cerebral cortex are con	cerned with			
		a)Purely sensory funct	ions	b)purely motor func	tions		
		c)Functions neither cle	early sensory nor motor	d)Vision and equilib	orium		

49.	Frontal, parietal, temp a)optic lobes	boral and occipital lobe b)Cerebellum	es are the parts of c)thalamencephalon	d)Cerebrum
50.	Anterior choroid plexu	ıs is present in		
	a)Roof of diencephalon	b)Roof of medulla	c)Floor of epithalamus	d)Floor of medulla
51.	Centers for satiety, fee	ding and thirst are pre	esent in	
	a)medulla oblongata	b)hypothalamus	c)pons varolii	d)Cerebellum
52.	Limbic system of brain	n does not include		
	a)Cerebral hemispheres	b)Amygdala	c)Arborvitae	d)Hippocampus
53.	Dorsal portion of mese	encephalon consists of		
	a)Posterior choroid plex	tus b)Corpora quadriger	mina c)Arborvitae	d)Corpus callosum
54.	Crus cerebrum connec	ets		
	a)Paracoels with diocoe	1	b)Cerebellum with spin	al cord
	c)Cerebral hemispheres	with pons	d)Cerebellar halves	
55.	'The tree of life' is			
	a)Medulla oblongata	b)Arbor vitae	c)Pons varolii	d)Organ of corti
56.	Pineal stalk arises from	n		
	a)Dorsal surface of epith	halamus	b)Ventral surface of ep	ithalamus
	c)Ventral surface of hyp	oothalamus	d)Dorsal surface f hypo	othalamus
57.	Control and coordinat	ion of muscular mover	nents is the function of	
	a)Mesencephalon	b)Hypothalamus	c)Cerebellum	d)Medulla
58	This one forms a bride	a hatwaan twa aarahal	lar homisnhoros	
30.	a)Corpus callosum	b)Pons varolii	c)Cerebral peduncles	d)Infundibulum
	, 1	,	, 1	,
59.	Relay station between	the cerebellum, spinal	cord and rest of the bra	ain is
	a)Medulla oblongata	b)Mesencephalon	c)Cerebellum	d)Pons varolii
60.	Brain stem does not in	clude		
	a)Medulla oblongata	b)Mesencephalon	c)Hippocampus	d)Pons varolii
61.	The vascular folded st	ructure seen in the me	dulla oblongata is	
	a)Arbor vitae	b)Anterior choroid plex	kus c)Posterior choroid p	olexus d) Infundibulum
62.	Iter lies in between			
	a)paracoel and diocoel	b)metacoel and paracoe	el c)Paracoel and parac	oel d)metacoel and diocoel
(2	N			
03.	of brain	elated to language lear	ning in right handed pe	copie is performed in this part
	a)left cerebral hemisphe	ere	b)Right cerebral hemis	phere
	c)left cerebellar half		d)Right cerebellar half	
64	Domovel of which read	t of brain sources imme	diata daath?	
V4.	a)Cerebrum	b)Optic lobes	c)Medulla oblongata	d)Olfactory lobes

	65.	Study the following a I. The midbrain, pons II. The inner parts of III. Each cerebral her Correct ones of the al	bout the brain. s, cerebellum and medu cerebral hemispheres, misphere is divided inte	ılla oblongata together : amygdala and hippoca o frontal, parietal, temp	form brain stem. mpus form the limbic system. ooral and occipital lobes.	
		a)Only I and III	b)Only I and II	c)Only II and III	d)I, II, III	
	66.	Osmoregulatory and a)Pons varolii	thermoregulatory cent b)Hypothalamus	ers are located in c)Medulla oblongata	d)Diencephalon	
	67.	Match the following a <u>BRAIN STRUCTUR</u> A)Superior colliculi-H B)Association areas C)Inferior colliculi-an D)Cerebellum	and choose the correct <u>E</u> nearing uditing	combination <u>RELATED FUNCTION</u> 1.Auditory 2.Memory, communion 3.Vison 4.relay station between 5. maintains equilibries	<u>ON</u> cation en spinal Cord and brain	
	68.	The correct answer is a)A-3, B-2, C-4, D-5 Match the following	b)A-3, B-2, C-4, D-1 and choose the correct	c)A-2, B-4, C-1, D-5	d)A-3, B-2, C-1, D-5	
hing.com		A)Pnemotaxic centre B)satiety centre C)Vomiting centre D)Arborvitae The correct answer is a)A-3 B-1 C-4 D-2	b)A-4 B-2 C-5 D-1	1.Medulla oblongata 2.Cerebellum 3.Hypothaalmus 4.Epithalamus 5.Pons varolii	d)A-2 B-3 C-1 D-5	
tcoac	69.	Photosensitive substa a)melanin	nce concerned with vis b)retinol	ion is c)rhodpsin	d)collagen	
٨٧	70.	In humans color visio a)rods	on is controlled by b)rhabdome	c)crystalline	d)cones	
	71.	This part of eyes is us a)Optic disc	seful much while walki b)fovea	ng,reading and driving c)blind spot	d)sclera	
	72.	Main function of vestibular apparatus is a)Processing information of vision c)Perceive the motion of head		b)perceive and transmits sound vibrations d)maintaining the shape of internal ear		
	73.	Lens and cornea are a a)vitreous humor	nourished by b)Aqueous humor	c)sclera	d)canal of schlemm	
	74.	Principle function of a)Filtering dust particle c)Formation of tears	iris is es	b) Preventing entry ofd)To regulate the amount	UV rays into eyes unt of light that enter the eyes	
6	75.	Stapes is attached to a)Fenestra ovalis	b)Eustachian tube	c)Fenestera rotunda	d)Tympanic membrane	

76.	Otolith organ compr a)Ampulla and crista	ises	b)Cochlea and semicir	cular canals
	c)sacculeand utricle		d)Ampulla and macula	l
77.	Set of non-vascular p	parts of eyes is		
	a)Lens,cornea	b)Ciliarybody,iris	c)Choroid,cornea	d)Retina,lens
78.	The correct pair with a)Optic disc-No rods b	h regard to human sense but only cones	e organs is b)Macula-vestibule d)Tympanic cavity-Flu	id filled cavity
		op5111	d) i ynipune cuvity i fe	
79.	Near or far vision in a)Position of retina	a normal person is attai b)Area of pupil	ned by changing the c)Shape of lens	d)Position of cornea
80.	Aqueous humor is se a)Lacrimal glands	b)Ciliary processes	c)Iris	d)Scleral venoussinus
81.	Receptors for gravity a)Cochlea	y are present on b)Macula	c)Semicicular canals	d)Middle car
82.	Organ of corti is loca a)Sacculus	a ted in b)Scala tympani	c)Scala media	d)Scala vestibule
83.	Eustachian tube com	nects		
	a)External ear and middle ear c)Middle ear and internal ear		b)Pharynx and middle ear d)Oral cavity and external ear	
84.	The correct order of	arrangement of ear ossi	cles from outer to inne	r is
	a)Malleus,stapes,incus	s b)Incus,stapes,malleus	c)Stapes,malleus,incus	d)Malleus,incus,stapes
85.	Aqueous and vitreou	s humors are separated	by	
	a) Conjunctiva	b) Cornea	c) Iris	d) Lens
86.	Eustachian tube help	os in		
	a)Carry impulses to internal ear c)Maintains body posture		b)Carry vibrations from external ear d)Maintain equilibrium on either side of ear drum	
87.	Canal of schlemm is present at the junction o a)Middle ear and pharynx		of b)Sclera and cornea of eye	
	c)Lens and retina of e	ye	d)Fovea and optic disc	,
88.	The layer closest to t a)Photoreceptor layer	he choroid coat of eyes is b)Bipolar cell layer	s c)Ganglion cell layer	d)Multipolar cell layer
89.	Visual purple is impo a)Colour vision	ortant in b)Scotopic vision	c)Photopic vision	d)Binocular vision
90.	The up-ward and do	wn-ward movement in a	a lift are perceived by	

01	Work dono in t	he given P V diegram in	the evalue process is	•
<i>7</i> 1.	work done in t	ne given i - v utagram m	the cyclic process is	, \mathbf{P} (2P,V) (2P,
	a) PV	b) 2PV		$(\mathbf{P} \mathbf{V})$ $(\mathbf{P} \mathbf{2V})$
	c) PV/2	d) 3PV		(\mathbf{r},\mathbf{v}) $(\mathbf{P},\mathbf{2v})$
92.	When a system i	is taken from state i to a sta	ate f along path iaf, Q =	= 50 J and W = 20 J. Along p
	ibf, Q = 35 J. If V	W = -13 J for the curved re	eturn path f I, Q for th	is path is PA
	a) 33 J	b) 23 J		
	c) -/ J	d) -43 J		i
93.	When a monoate	omic gas expands at consta	ant pressure, the perce	ntage of heat supplied that
	increases the int	ernal energy of the gas and	l that which is involved	d in expansion is
	a) 75%, 25%	b) 25%, 75%	c) 60%, 40%	d) 40%, 60%
94.	In the following	indicator diagram, the net	amount of work done	will be P
	a) Positive	b) Negative		$\begin{pmatrix} 1 \end{pmatrix}$ 2
	c) Zero	d) Infinity		$\langle \gamma \rangle$
95.	Efficiency of a C	Carnot engine is 50% when	temperature of outlet	is 500 K. In order to increas
	efficiency up to	60% keeping temperature	of intake the same wha	at is temperature of outlet
	a) $200 V$	1) 400 17		
	a) 200 K	b) 400 K	c) 600 K	d) 800 K
96.	a) 200 K If the door of ref	b) 400 K frigerator is kept open, the	c) 600 K n which of the followin	d) 800 K ng is true
96.	a) 200 K If the door of rel a) Room is cooled	b) 400 K frigerator is kept open, the d	c) 600 K on which of the followin b) Room is heated	d) 800 K ng is true
96.	a) 200 K If the door of rel a) Room is cooled c) Room is either	b) 400 K frigerator is kept open, the d · cooled or heated	 c) 600 K en which of the followin b) Room is heated d) Room is neither of 	d) 800 K ng is true cooled nor heated
96. 97.	a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiab	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of	 c) 600 K m which of the followin b) Room is heated d) Room is neither of work is done on 2 mol 	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to
96. 97.	a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiab reduce its volum	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of he by 50%. The change in i	 c) 600 K m which of the followin b) Room is heated d) Room is neither of work is done on 2 molts temperature is near 	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly (R = 8.3 JK ⁻¹ mol ⁻¹)
96. 97.	 a) 200 K If the door of refailer a) Room is coolea c) Room is either During an adiab reduce its volum a) 40 K 	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of he by 50%. The change in i b) 33 K	 c) 600 K m which of the followin b) Room is heated d) Room is neither of work is done on 2 mol ts temperature is near c) 20 K 	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly (R = 8.3 JK ⁻¹ mol ⁻¹) d) 14 K
96. 97. 98.	 a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiab reduce its volum a) 40 K The r.m.s. veloci molecule at 127° 	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of he by 50%. The change in i b) 33 K ity of oxygen molecular at C is	 c) 600 K en which of the followin b) Room is heated d) Room is neither of work is done on 2 mole ts temperature is nearling c) 20 K 16°C is 474 m/sec. The 	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly (R = 8.3 JK ⁻¹ mol ⁻¹) d) 14 K e r.m.s velocity in m/s of hyd
96. 97. 98.	 a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiab reduce its volum a) 40 K The r.m.s. veloci molecule at 127° a) 1603 	b) 400 K frigerator is kept open, the d cooled or heated oatic compression, 830 J of he by 50%. The change in i b) 33 K ity of oxygen molecular at C is b) 1896	 c) 600 K m which of the followin b) Room is heated d) Room is neither of work is done on 2 mol ts temperature is nearl c) 20 K 16°C is 474 m/sec. The c) 2230.59 	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly (R = 8.3 JK ⁻¹ mol ⁻¹) d) 14 K e r.m.s velocity in m/s of hyd d) 2730
96. 97. 98. 99.	 a) 200 K If the door of ref a) Room is coolea c) Room is either During an adiab reduce its volum a) 40 K The r.m.s. veloci molecule at 127° a) 1603 At 0°K which of 	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of he by 50%. The change in i b) 33 K ity of oxygen molecular at C is b) 1896 the following properties of	 c) 600 K m which of the followin b) Room is heated d) Room is neither of work is done on 2 molest to the the the the the the the the the the	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly (R = 8.3 JK ⁻¹ mol ⁻¹) d) 14 K e r.m.s velocity in m/s of hyd d) 2730
96. 97. 98. 99.	 a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiab reduce its volum a) 40 K The r.m.s. veloci molecule at 127° a) 1603 At 0°K which of a) Kinetic energy 	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of he by 50%. The change in i b) 33 K ity of oxygen molecular at C is b) 1896 the following properties of b) Potential energy	 c) 600 K m which of the followin b) Room is heated d) Room is neither of work is done on 2 molestic temperature is nearlinedim c) 20 K 16°C is 474 m/sec. The c) 2230.59 f a gas will be zero? c) Vibrational energy 	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly ($\mathbf{R} = \mathbf{8.3 \ JK^{-1} \ mol^{-1}}$) d) 14 K e r.m.s velocity in m/s of hyd d) 2730 gy d) Density
96. 97. 98. 99.	 a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiab reduce its volum a) 40 K The r.m.s. veloci molecule at 127° a) 1603 At 0°K which of a) Kinetic energy The perfect gas of 	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of he by 50%. The change in i b) 33 K ity of oxygen molecular at C is b) 1896 the following properties of b) Potential energy equation for 4 gram of hyd	 c) 600 K m which of the followin b) Room is heated d) Room is neither of work is done on 2 molestic temperature is nearling c) 20 K 16°C is 474 m/sec. The c) 2230.59 f a gas will be zero? c) Vibrational energe 	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly (R = 8.3 JK ⁻¹ mol ⁻¹) d) 14 K e r.m.s velocity in m/s of hyd d) 2730 gy d) Density
96. 97. 98. 99. 100.	 a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiab reduce its volum a) 40 K The r.m.s. velocimolecule at 127° a) 1603 At 0°K which of a) Kinetic energy The perfect gas of a)PV = RT 	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of ne by 50%. The change in i b) 33 K ity of oxygen molecular at C is b) 1896 the following properties of b) Potential energy equation for 4 gram of hyd b)PV = 2RT	c) 600 K en which of the followin b) Room is heated d) Room is neither of work is done on 2 mol ts temperature is nearly c) 20 K 16°C is 474 m/sec. The c) 2230.59 f a gas will be zero? c) Vibrational energy lrogen gas is c) PV = $\frac{1}{2}$ RT	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly ($\mathbf{R} = \mathbf{8.3 \ JK^{-1} \ mol^{-1}}$) d) 14 K e r.m.s velocity in m/s of hyd d) 2730 gy d) Density d)PV= 4RT
96. 97. 98. 99. 100.	 a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiab reduce its volum a) 40 K The r.m.s. veloci molecule at 127° a) 1603 At 0°K which of a) Kinetic energy The perfect gas of a)PV = RT The number of content 	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of the by 50%. The change in i b) 33 K ity of oxygen molecular at 'C is b) 1896 the following properties of b) Potential energy equation for 4 gram of hyd b)PV = 2RT degrees of freedom for eacl	c) 600 K on which of the followin b) Room is heated d) Room is neither of work is done on 2 mol ts temperature is nearly c) 20 K 16°C is 474 m/sec. The c) 2230.59 f a gas will be zero? c) Vibrational energy lrogen gas is c) PV = $\frac{1}{2}$ RT h atom of a monoatomic	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly (R = 8.3 JK ⁻¹ mol ⁻¹) d) 14 K e r.m.s velocity in m/s of hyd d) 2730 gy d) Density d)PV= 4RT ic gas is
96. 97. 98. 99. 100. 101.	 a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiab reduce its volum a) 40 K The r.m.s. veloci molecule at 127° a) 1603 At 0°K which of a) Kinetic energy The perfect gas of a) PV = RT The number of of a) 3 	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of the by 50%. The change in i b) 33 K ity of oxygen molecular at 'C is b) 1896 the following properties of b) Potential energy equation for 4 gram of hyd b)PV = 2RT degrees of freedom for each b) 5	c) 600 K on which of the followin b) Room is heated d) Room is neither of work is done on 2 mol ts temperature is nearly c) 20 K 16°C is 474 m/sec. The c) 2230.59 f a gas will be zero? c) Vibrational energy lrogen gas is c) PV = $\frac{1}{2}$ RT h atom of a monoatomic c) 6	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly (R = 8.3 JK ⁻¹ mol ⁻¹) d) 14 K e r.m.s velocity in m/s of hyd d) 2730 gy d) Density d)PV= 4RT ic gas is d)1
96. 97. 98. 99. 100. 101. 102.	 a) 200 K If the door of ref a) Room is cooled c) Room is either During an adiabireduce its volum a) 40 K The r.m.s. velocimolecule at 127° a) 1603 At 0°K which of a) Kinetic energy The perfect gas of a) PV = RT The number of of a) 3 In an adiabatic energy 	b) 400 K frigerator is kept open, the d cooled or heated patic compression, 830 J of he by 50%. The change in i b) 33 K ity of oxygen molecular at C is b) 1896 the following properties of b) Potential energy equation for 4 gram of hyd b)PV = 2RT degrees of freedom for each b) 5 expansion of a gas initial an	c) 600 K on which of the followin b) Room is heated d) Room is neither of work is done on 2 mol ts temperature is nearly c) 20 K 16°C is 474 m/sec. The c) 2230.59 f a gas will be zero? c) Vibrational energy lrogen gas is c) PV = $\frac{1}{2}$ RT h atom of a monoatomic c) 6 nd final temperatures a	d) 800 K ng is true cooled nor heated les of a diatomic ideal gas to ly ($\mathbf{R} = \mathbf{8.3 \ JK^{-1} \ mol^{-1}}$) d) 14 K e r.m.s velocity in m/s of hyd d) 2730 gy d) Density d)PV= 4RT ic gas is d)1 are T_1 and T_2 respectively, th

a) $\frac{R}{\gamma - 1}(T_2 - T_1)$ b) $\frac{R}{\gamma - 1}(T_1 - T_2)$ c) $R(T_1 - T_2)$ d)Zero

	103.	An ideal gas is take	n around ABCA as s	shown in the above P-V dia	agram. The work done during	g a	
		cycle is			- 1		
		a) 2PV	b) PV		^P [^B (3P, 3V)		
		c) 1/2PV	d)Zero				
		•) 1/=1 /	u)2010				
					$(P,V) \qquad (P,3V) \qquad \qquad$		
	104	The near estime an es	da of firm molecules	one 2151(1(and 12 hr	E = D	۰d	
	104.	in km/sec will be	as of five molecules	are 2,1.5,1.0,1.0 and 1.2 km	n/sec. The most probable spe	ea	
		a) 2	b) 1.58	c) 1.6	d) 1.31		
	105.	In which of the follo	owing processes the i	internal energy of the syste	em remains constant?		
		a) Adiabatic	b) isochoric	c) Isobaric	d) Isothermal		
		,	,				
	106.	A given mass of a g	as expands from the	e state A to the state B by t	three paths 1,2 and 3 as show	n in	
		the figure. If W_1, W_2	and W_3 respectivel	y by the work done by the	gas along the three paths the	n	
		a) $W_1 > W_2 > W_3$			А		
		b) $W_1 < W_2 < W_3$					
		c) $W_1 = W_2 = W_3$					
		d) $W_1 < W_2 = W_3$			V		
	107.	On compressing a g	vas suddenly, its tem	nerature			
	1071	a) increases	b) decreases	c) remains constant	d) all the above		
		a) mereases	D) uccreases	c) remains constant	d) all the above		
ing.c	108.	Two hollow spheres of same material one with double the radius of the other and double the thickness of the other filled with ice, The ratio of times in which ice gets melted in the two spheres is					
chi		a) 2:1	b) 1:2	c) 4:1	d) 1:4		
vvtcoa	109.	Two spheres A and rise in their temper fall is a) 3:1	B with masses in th catures on reaching b) 1:3	ne ratio 2:3 and specific he the ground are in the rati c) 4:3	eat 2:3 fall freely from rest. In o 1:2 the ratio of their heigh d) 3:4	f the ts of	
	110.	A 50kg man is run	ning at a speed of 1	$8kmh^{-1}$. If all the kinetic of	energy of the man can be use	ed to	
		increase the temper	rature of water from	$1 \ 20^{\circ}C$ to $30^{\circ}C$, how mu	ich water can be heated with	this	
		energy?					
		a) 15 g	b) 20 g	c) 30 g	d) 40 g		
	111.	Find the external w	ork done by the sys	tem in kcal, when 20 kcal	of heat is supplied to the sys	stem	
		and the increase in	the internal energy i	is 8400 J (J=4200J/kcal)			
		a) 16 kcal	b) 18 kcal	c) 20 kcal	d) 19 kcal		
	110						
	112.	Air expands from 5	ntres to 10 ntres at	2 atm pressure. External v	vorkdone is		
		a) 10J	b) 1000J	c) 3000J	d) 300J		
	112	The temperature of	f 5 moles of a gas a	t constant volumo is abor	$100^{\circ}C$ to $120^{\circ}C$	The	
	113.	change in internal	energy is 80J. The t	otal heat capacity of the g	gas at constant volume will h	be in	
		joule/Kelvin is					
9		a) 8	b) 4	c) 0.8	d) 0.4		

114	A Staal hall of mage		m a haisht af 10m and	d hourse to a height of 5 American
114.	A Steel Dall of mass	dissipated onergy in	this process is about	rbod by the ball the rise in its
	temperature is (spec	vificheatofsteel 460 ik	$a \sigma^{-1} k^{-1} \sigma = 19 m s^{-2}$	The by the ban, the fise in its
	a) $0.01^{\circ}C$	b) $0.1^{\circ}C$	c) $1^{0}C$	d) $1 \ 1^0 C$
	u) 0.01 C	c) 0.1 C	0) 1 0	
115.	Cooking is difficult o	on mountains because		
	a) water boils at low to	emperature	b) water boils at hig	gh temperature
	c) water does not boll		d) it is cool there	
116.	A large block of ice i	s placed on a table wh	en the surroundings a	re at $0^{\circ}C$
	a)ice melts at the side	s b) ice melts at the top	c) ice melts at the b	ottom d) ice does not melt at all
117.	The ratio of densities capacities per unit ve	s of two substances is plume is	2 :3 and that of specifi	ic heats 1 : 2. The ratio of thermal
	a) 1 : 2	b) 2 : 1	c) 1 : 3	d) 3 : 1
110	T I' 'I A ID	e 1 1 1		
118.	thermal capacity, the	of equal volumes hav en the ratio of their de	e their specific heats i nsities is	in the ratio 2 : 3. If they have same
	a) 1 : 1	b) 2 : 3	c) 3 : 2	d) 5 : 6
			0	
119.	Specific heat of alun	ninium is $0.25 cal / g$ -	$-^{\circ}c$. The water equiv	valent of an aluminium is vessel of
	mass one kilogram is		250 - 1/0	
	a) 40 <i>cal</i> /° c	b) 250 <i>g</i>	c) 250 <i>cal</i> /° c	d) 40 <i>g</i>
120.	Two liquids A and I	B are at $30^{\circ}c$ and 20	^{0}c respectively. Whe	n they are mixed in equal masses
	the temperature of the	ne mixture is found to	be $26^{\circ}c$. The ratio of	specific heats is
	a) 4 : 3	b) 3 : 4	c) 2 : 3	d) 3 : 2
101	If 10g of the ice of	0^0 a is mixed with 10	a of water at 100° a	then the final temperature of the
141.	mixture will be	0 c is mixed with 10	g of water at 100 C,	then the final temperature of the
	a) $5^{0}c$	b) 10 [°] c	c) 100 <i>K</i>	d) $0^{0}c$
122.	The heat energy req	uired to vapourise 5 k	g of water at 373 K is	
	a) 2700 K. cal	b) 1000 K.cal	c) 27 K.cal	d) 270 K.cal
123.	A metal block absorl	bs 4500 cal of heat who	en heated from $30^{\circ}c$	to $80^{0}c$. Its thermal capacity is
	a) 90 <i>gm</i>	b) 90 <i>cal</i> / ⁰ <i>C</i>	c) 9 <i>gm</i>	d)) 9 <i>cal</i> / ⁰ <i>C</i>
104			00 (00 1 9)	20°
124.	Inree liquids A, B	and C of masses 4	uugm, buugm and 80	$JUgm \text{ are at } 30 \ c,40 \ c \ and \ 50 \ c$
	respectively. When	A and B are mixed	d resultant temperati	ure is $36^{\circ}c$ are mixed resultant
	temperature is $44^{\circ}c$	Then ratio of their sp $b \cdot 3 \cdot 2 \cdot 1$	ecific heats are $(2 \cdot 2 \cdot 1)$	d) $1 \cdot 4 \cdot 9$
	a <i>j 4</i> . 1 . 1	0.3.2.1	0,2.2.1	u) 1. 4 . 7
125.	50g of copper is heat	ted to increase its tem	perature by $10^{0}C$. If	the same quantity of heat is given
	to 10 g of water, the	rise in its temperatur	e is	
	$(S_{cu} = 420 J / kg / ^0 G)$	$C and S_w = 4200 J / kg$	$(1 - 1)^{0} C$	
	a) $5^{0}C$	b) $6^{0}C$	c) $7^{0}C$	d) $8^{0}C$

	126.	Power of man who c	an chew 0.3 kg ice in or	ne minute is (in cal/s)	
	1200	a) 400	b) 4	3) 24	d) 240
\geq	127.	One mole of oxygen i supplied to the gas to	is heated at constant p double its volume (R i	ressure starting at $0^0 C$ is the molar gas constan	. The heat energy that must be at)
		a) $2.5 \times 273 \times R$	b) 3.5×273× <i>R</i>	c) $2.5 \times 546 \times R$	d) $3.5 \times 546 \times R$
	128.	A carnot's engine w should the temperatu	hose sink is at a tempe 1re of the source be inc	rature of 300K has an reased so as to increase	efficiency of 40%.By how much the efficiency to 60%
		a) 250K	b) 275 K	c) 300 K	d) 325 K
	129.	An ideal Carnot's en then the temperature	gine whose efficiency is e of sink will be	s 40% receives heat at 5	500K. If the efficiency to be 50%
		a) 600 K	b) 800 K	c) 1000K	a) 250 K
	130.	The average kinetic	energy of a molecule of $\sqrt{\frac{1}{T}}$	f a gas at absolute temp	erature T is proportional to T^2
		a) 1/ <i>1</i>	b) \sqrt{I}	c) <i>I</i>	d) <i>T</i> ²
	131.	At a given temperatu	tre if V_{rms} is the root m	ean square velocity of t	he molecules of a gas and V_s be
		the velocity of sound	in it, then these are rel	ated as $\left(\gamma = \frac{C_P}{C_V}\right)$	
0m		a) $v = v_c$	b) $v = \sqrt{\frac{3}{2}} \times v_c$	c) $v = \sqrt{\frac{\gamma}{\gamma}} \times v_c$	d) $y = \sqrt{\frac{3}{2}} \times y_a$
ວ ຄ		ery ms S	$\gamma \gamma $	$\gamma \gamma $	$\gamma \gamma $
chin	132.	The average degrees expands at constant j a) 75J	of freedom per molecu pressure. The heat abso b) 100 J	lle for a gas is 6. The ga orbed by gas is c) 150 J	s performs 25J of work when it d) 125 J
C08	100				
vt	133.	of the molecules of th	The velocity, V_{rms} the averages are in the order :	erage velocity V_{av} and t	he most probable velocity, V_{mp}
		a) $v_{mp} > v_{avg} > v_{rms}$	b) $v_{rms} > v_{avg} > v_{mp}$	c) $v_{avg} > v_{mp} > v_{rms}$	d) $v_{mp} > v_{rms} > v_{avg}$
	134.	If number of molecu	les of H_2 are double the	nan that of O_2 , then rate	tio of kinetic energy of hydrogen
		and that of oxygen at	t 300 <i>K</i> is		
		a) 1: 1	b) 1 : 2	c) 2 : 1	d) 1 : 16
	135.	Air expands from 5 l a) 10J	itres to 10 litres at 2 at b) 1000J	m pressure. External w c) 3000 J	orkdone is d) 300 J
			CHEMI	STRY	
	136.	0.45 gm of an organi	c compound containing	g only carbon, hydroger	n and nitrogen on combustion
		a) $C_4 H_3 N_2$	b) $C_3H_4N_2$	c) $C_4 H_3 N$	d) $C_3 H_4 N$
11					

	137.				
>		H_3C $CH = C$	$=CH_2$ is		
		H CH C	CH_{3}		
		a) E-isomer	b) Z-isomer	c) Cis-isomer	d) Trans-isomer
	138.	The number of σ an	d π bonds in a molecul	e of acetonitrile are r	respectively
		a) 2,5	b) 3,4	c) 4,3	d) 5,2
	139.	The optical inactivity a) dextro-tartaric acid	y due to internal compe b) laevo-tartaric acid	nsation can be exhibi c) racemic-tartaric a	ted by: acid d) meso-tartaric acid
	140.	The structure of ally	lchoride is		
		a) $CH_2 = CH - CH_2$	$_{2}Cl$	b) $CH_2 = C Cl -$	$-CH_3$
		c) $CH Cl = CH -$	CH ₃	d) $CH Cl = C C$	$Cl - CH_3$
	141	Which of the follow:	na ia katana analia aana	o	
	141.	a) Anthracene	b) Pyrrole	c) Phenol	d) Isobutylene
	142.	Distillation is used to	separate liquids which	differ in the b.p by	
B		a) $5^{0}C$	b) $10^{0}C$	c) $30 - 80^{\circ}C$	d) $100^{\circ}C$
[] []	143.	Example of chain iso	merism are		
5		a) Ortho and meta tolu	ic acids	b) Methyl acetate ar	nd ethyl formate
hir		c) Pentanoic acid and	2-methyl butanoic acid	d) 2-pentanone and	3-pentanone
acl	144.	How many stereoison	mers does this molecule	have	
LCO		$cH_3 cH = cH cH_2 c$	b) 6	c) 8	d) 2
	145.	In the hydrocarbon	$CH_3 - CH = CH - CH - CH - CH - CH - CH - CH$	$-CH_2 - C \equiv CH$	
		The state of hybrizat	ion of carbons 1.3 and 5	5 2 1 5 are in the following	sequence:
		a) sp^2 , sp , sp^3	b) sp, sp^3, sp^2	c) sp, sp^2, sp^3	d) sp^3 , sp^2 , sp
	146.	Which of the followi a) Butanol	ng compounds will exhi b) 2-Butyne	bit cis-trans (geometr c) 2-Butenol	rical) isomerism? d) 2-Butene
	147.	The correct decreasi IUPAC system of no a) $-COOH$, $-SO_2H$	ng order of priority for menclature is $T = CONH_{23}$, -CHO	the functional group b) $-SO_2H$, $-COO$	s of organic compounds in the DH_{2} , $-CONH_{2}$, $-CHO$
		c) <i>-CHO</i> , <i>-COOH</i> ,	$-SO_3H, -CONH_2$	d) $-CONH_2, -CH_2$	$HO, -SO_3H, -COOH$
12	148.	The IUPAC name of a) 4-Bromo-3-cyanop b) 2-Bromo-5-hydroxy c) 2-Cyano-4-hydroxy d) 6 Bromo 2 hydroxy	the following compount henol ybenzonitrile ybromobenzene	d is	
		u) o-bromo-3-nyarox	ydenzonnne	Br	

		nitrogen collected at	300 K temperature and	725 mm pressure. l	If the aqueous tension at 300 K
		25 mm, the percenta	b) 16.76	npound is:	d) 17 36
		a) 16.20	0) 10.70	c) 15.70	u) 17.30
	150.	The number of isom	ers of the compound C_2	BrFCl is	
		a) 3	b) 4	c) 5	d) 6
	151.	Correct order of state a) cis-2-butene.1-bute c) 1-butene>cis-2-but	bility is: ene>trans-2-butene ene>trans-2-butene	b) trans-2-butene> d) cis-2-butene>tra	cis-2-butene>1-butene ns-2-butene>1-butene
	152.	Which of the followi	ng does not exhibit cis-t	rans isomerism	
		a) $C_6H_5CH = NOH$	<i>I</i> b) $C_6H_5N = NC_6H_5$	c)	d) None of these
	153.	0.37 g of a given con a) 52.3%	b) 72.6%	he silver bromide, th c) 80.2%	he percentage of bromine in it d) 28.4%
	154.	During hearing of a carried out. He aske According to you wi	court case, the judge sus d the forensic departmen nich technique can give t	spected that some ch nt to check the ink u he best results?	nanges in the documents had l used at two different places.
_		a) Column chromatog	graphy	b) Solvent extraction	on
		c) Distillation		d) Thin layer chror	natography
5	155.	The principle involv	ed in paper chromatogra	aphy is	
		a) adsorption	b) partition	c) solubility	d) volatility
CE				/	
03	156.	What is the hybridis	ation state of benzyl car	bonium ion $\langle \ \subset$	CH_2^+
2					
		a) sp^3	b) sp^2	c) spd^2	d) sp^2d
	157.	Nitrogen detection in	n an organic compound i	is carried out by La	ssaigne's test. The blue colour
		formed corresponds	to which of the following	g formulae	
		a) $Fe_3[Fe\ CN_6]_3$	b) $Fe_3 [Fe CN_6]_2$	c) $Fe_4[Fe\ CN_6]$	d) $Fe_4 [Fe CN_6]_2$
	158.	In the Kjeldahl's me	ethod for estimation of n	itrogen present in a	soil sample, ammonia evolved
		from 0.75 g of samp	le neutralized 10 mL of 1	$M H_2 SO_4$. The per	rcentage of nitrogen in the soi
		a) 37.33	b) 45.33	c) 35.33	d) 43.33
	159.	$CH_3CH_2CH_2OH$ is	s a functional isomers of		
		a) $C_2H_5OCH_3$	b) $CH_3OC_3H_7$	c) $CH_3CH_2CH_2C$	OCH_2CH_3 d) CH_3CHOHC_2
	160.	The IUPAC name of	$CH_3 - C \equiv C - CH C$	H_{3} , is	
	-	a) 4-methyl-2-nentyn	e e	b) 4 4-dimethyl_?_l	outvne
		$a_j \neq \text{methyl}^{-2}$ -pentyl	~	3) = -1	Jacque



>

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72. The most suitable method for separation of a 1:1 mixture of ortho and para nitrophenols isa) sublimationb) chromatographyc) crystallizationd) steam distillation

173. Which of the following is optically active

a)
$$CH_3 - CH_3 - CH_3$$

174. The correct IUPAC name for
a) 5-methyl 4- 1,2-dimethylpropyl heptane
b) 3-methyl 4- 1,2 - dimethylpropyl heptane
c) 2,3,5-trimethyl 4-propylheptane
d) 4-propyal-2,3,5-trimethylpeptane
d) 4-propyal-2,3,5-trimethylpeptane
175. Total number of stereo isomers (optical and geometrical) of the following compound are

$$CHCl = CH - CH CH_1 C_2H_1$$

a) 2 b) 4 c) 6 d) 8
176. The number of ether isomers possible for $C_4H_{10}O$ are
a) 2 b) 5 c) 4 d) 3
177. Keto-enol tautomerism is observed in:
a) $C_5H_6 - \overset{f}{C} - CH_2 - \overset{f}{C} - C_8H_5$ b) $H_5C_6 - \overset{f}{C} - CH_3$
c) $H_5C_6 - \overset{f}{C} - H$ d) Both (a) and (b)
178. The best method for the separation of naphthalene and benzoic acid from their mixture is
a) distillation b) sublimation c) chromatography d) crystallization
179. If a compound has a asymmetric carbon atoms, then maximum number of optical isomers are
given by the formula
a) 2^n b) $\left(\frac{1}{2}\right)^n$ c) $\sqrt{2n}$ d) $2\sqrt{n}$
180. The correct statement about the compound (A), (B) and (C) is:
 $\stackrel{f}{=} \underbrace{\overset{OOH}{=}}_{OOH}$ $\stackrel{f}{=} \underbrace{\overset{OOH}{=}}_{OOCH_1}$
a) (A) and (B) are identical
b) (B) and (C) are enantiomers
d) (A) and (B) are enantiomers

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Test	Date	BOTANY	Z00L0GY	PHYSICS	CHEMISTRY
TEST - 1	2/9/2018	Morphology of Flowering Plants (Root, Stem, Leaf, Flower, Inflorescence)	The Living World	Physical World, Units and Measurements,	Structure of Atom
TEST - 3	16/09/2018	Morphology of Flowering Plants (Fruits & Taxonomy)	Animal Kingdom - Non Chordates	Motion in a Straight Line	Classification of Elements and Periodicity in Properties
TEST - 5	30/09/2018	Anatomy of Flowering Plants	Animal Kingdom - Chordates - Fishes, Amphibia, Reptilia, Aves & Mammalia	Motion in a Plane	Chemical Bonding and Molecular Structure
TEST - 7	14/10/2018	Cell: The Unit of Life	Animal Kingdom - Earth Worm, Cockroach, Frog	Laws of Motion	States of Matter
TEST - 9	28/10/2018	Biomolecules, Cell Cycle and Cell Division	Structural Organisation	Work, Energy and Power	Thermodynamics
TEST - 11	11/11/2018	Plant Physiology- Transport in Plants	Human Physiology: Digestion and Absorption	System of Particles and Rotational Motion	Chemical Equilibrium and Ionic Equilibrium
TEST - 13	25/11/2018	Mineral Nutrition	Breathing and Exchange of Gases	Gravitation	Some Basic Concepts of Chemistry, Redox Reactions
TEST - 15	9/12/2018	Photosynthesis in Higher Plants	Body Fluids and Circulation	Mechanical Properties of Solids	Hydrogen & Its Compounds
TEST - 17	23/12/2018	Respiration in Plants	Elimination of Excretory Products	Mechanical Properties of Fluids	The S-Block Elements (IA & IIA Group Elements)
TEST - 19	6/1/2019	Plant Growth and Development	Locomotion and Movement	Thermal Properties of Matter	The P-Block Elements (IIIA & IVA Group Elements)
TEST - 21	20/1/2019	Biological Classification	Neural Control and Co-ordination	Thermodynamics, Kinetic Theory,	Organic Chemistry – Some Basic Principles and Techniques
TEST - 23	3/2/2019	Plant Kingdom	Chemical Co-ordination and Integration	Oscillations, Waves	Hydrocarbons Alkanes, Alkynes & Benzene

Test	Date	BOTANY	Z00L0GY	PHYSICS	CHEMISTRY
TEST - 2	9/9/2018	Reproduction in Organisms - Plants	Human Reproductive System	Electric Charges and Fields,	The Solid State
TEST - 4	23/9/2018	Sexual Reproduction in Flowering Plants	Reproductive Health	Electrostatic Potential and Capacitance	Dilute Solutions
TEST - 6	7/10/2018	Sexual Reproduction in Flowering Plants	Principles of Inheritance and Variation	Current Electricity	Electrochemistry
TEST - 8	21/10/2018	Principles of Inheritance and Variation	Molecular Basis of Inheritance	Moving Charges and Magnetism	Chemical Kinetics
TEST - 10	4/11/2018	Principles of Inheritance and Variation	Evolution	Magnetism and Matter	Surface Chemistry
TEST - 12	18/11/2018	Molecular Basis of Inheritance	Human Health and Diseases	Electromagnetic Induction, Alternating Current	General Principles and Processes of Isolation of Elements
TEST - 14	2/12/2018	Molecular Basis of Inheritance	Strategies for Enhancement in Food Production	Electromagnetic Waves, Communication Systems	The P-Block Elements
TEST - 16	16/12/2018	Strategies of Enhancement in Food Production	Biotechnology and its Applications	Ray Optics and Optical Instruments	The D-And F-Block Elements Coordination Compounds
TEST - 18	30/12/2018	Microbes in Human Welfare	Organisms and Population	Wave Optics	Haloalkanes and Arenes, Alcohols, Phenols and Ethers
TEST - 20	13/1/2019	Biotechnology - Principles and Processes	Eco System	Dual Nature of Radiation and Matter	Aldehydes, Ketones and Carboxylic Acids
TEST - 22	27/1/2019	Biotechnological Applications in Medicine	Biodiversity and Conservation	Atoms, Nuclei	Amines, Chemistry in Everyday Life
TEST - 24	10/2/2019	Plant Ecology	Environmental Issues	Semiconductor Electronics: Materials, Devices and Simple Circuits	Biomolecules, Polymers



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